	the module/subject			Code			
Field of a		anic Compounds Synthes	sis Code 101070221101		211010720081		
Cher	^{study} nical Technolog	y	Profile of study (general academic, practica (brak)		Semester		
	path/specialty		Subject offered in:	Course	e (compulsory, elective)		
	Orga	anic Technology	Polish		obligatory		
Cycle of	study:		Form of study (full-time,part-time	e)			
Second-cycle studies			full-time				
No. of he	ours			No. of	credits		
Lectur	e: 2 Classes	s: 1 Laboratory: 3	Project/seminars:	-	5		
Status o	-	program (Basic, major, other) (brak)	(university-wide, from anothe	r field) (brak)			
Educatio	on areas and fields of sci	ence and art		ECTS and %)	distribution (number)		
dr ha ema tel. 6 Facu	onsible for subje ab. inż. Aleksandra Bo il: aleksandra.borowia 516653689 ulty of Chemical Techi Piotrowo 3 60-965 Poz	prowiak-Resterna ak-resterna@put.poznan.pl nology					
Prere	quisites in term	s of knowledge, skills and	d social competencies	6:			
1	Knowledge	1. The student has structured, the chemistry obtained during studie characteristic, basic reactions of	es in chemical technology first	degree, kno	egree, knows the		
		2. The student has a basic knowledge of information technology.					
2	Skills	 The student is able to solve th knowledge. 	e basic problem tasks of org	anic chemistr	c chemistry based on his		
		2. The student has the ability to obtain information from the identified sources.					
3	Social	1. The student is aware of the need to supplement his knowledge and its continuous updating.					
	competencies	2. The student knows the basic p	principles of teamwork.				
Assu	mptions and obj	ectives of the course:					
choose	the most optimal met	nsive and solid knowledge of orgation of obtaining the selected grouted and the selected grouted groutegrouted grouted grouted grouted groutegrouted groutegr	ups of organic compounds.				
	e recipe preparative of	o solve basic problems in the one- f selected organic compound base					
		ility to use molecular modeling to plex organic molecules as well as			hysicochemical		
4. Dev∉		ess of the responsibility for their fu					
	Study outco	mes and reference to the	educational results for	or a field o	of study		
Know	/ledge:						
		ded, solid knowledge of modern m taining the selected groups of orga		methods of p	lanning and selectior		
		knowledge of new, versatile and e					
		dern methods of study of the strue n of raw materials and products of			n molecular modeling		
useful f Skills							

1. The student can reliable select source of chemical and environmental information, critically evaluate the information obtained from the literature and electronic databases and carry out their analysis and draw conclusions. - [K_U01]

2. The student based on source material (also in English) and his own research can make a professional presentation of studied scientific issue. - [K_U06]

3. The student is able to use professional software, using them for the design of chemical processes and intermolecular interactions. - [K_U07]

4. The student knows and respects the principles of work-related health and safety in the chemical laboratory. - [K_U18]

Social competencies:

1. The student understands the need for continuous professional development. - [K_K01]

2. The student understands the importance of the responsibility that rests on all members of the team performing the task assigned; is aware of the need to respect the principles of teamwork. - [K_K04]

Assessment methods of study outcomes

Lectures - Assessment of knowledge and skills on the basis of the written exam (6 problem tasks based on course description of lectures).

Classes - Assessment prepared by the student (based on reference literature) presentation discusses the selected type of modern chemical reactions and to assess the knowledge he has acquired on the basis of test summary. Laboratory:

1) Synthesis of the product - oral answer before the preparation, analyzing how to execute of the planned synthesis (based on the research literature); evaluation of the practical implementation of the synthesis of the desired product; evaluation report containing an analysis of research literature and discuss the progress and result of the synthesis of an organic compound.

2) Molecular modeling - final test.

Course description

Lectures:

Factors influencing the reactivity of molecules and chemical reactions direction: types of interactions between molecules, acidity and basicity of organic compounds. electrophiles and nucleophiles. hard and soft acids and bases - sort by Pearson. Hammett equation, the role of the catalyst and solvent. Classification of solvents. Empirical parameters of solvent polarity. Phase transfer catalysis. Chemo-, regio- and stereoselective reactions. Stereospecific reactions. Selectivity and mechanisms of organic reactions: direct and conjugate addition (?1,2-addition? and ?1,4-addition?), elimination, electrophilic and nucleophilic substitution, reactions of oxidation and reduction. Protection of functional groups. Retrosynthetic analysis. Selective formation of carbon-carbon and carbon-heteroatom bonds, for example a metathesis reaction of alkenes, aldol condensation, Michael addition, Robinson annulation, coupling reactions with the participation of organometallic catalysts, pericyclic reactions. The use of organosilicone compounds, and organo-boron(III) compounds in organic synthesis. Examples of multi-step syntheses of organic compounds present in the environment (total syntheses).

Learning to use the primary literature. The discussion of the subject areas, with the active participation of students, some types of modern chemical reactions used in the multi-stage syntheses, among other compounds present in the environment. Laboratory:

1) Carrying out research literature on methods of obtaining the selected organic compound. Synthesis of the preparation using professional equipment and advanced purified techniques. Before making the preparation - a reminder and update industrial safety rules in force in the organic chemistry lab.

2) The introduction of the basic principles of molecular modeling - spatial manipulation of molecules with certain models of structural parameters in two and three dimensions, the basic techniques of molecular structures, modeling and measurement of structural parameters, building of multi-functional molecules, minimizing the energy of the molecule or molecules in the vacuum system.

Basic bibliography:

1. McMurry J., Chemia organiczna, PWN, Warszawa 2007.

2. Clayden J., Greeves N., Warren S., Wothers P., Chemia organiczna, tom I, II i III, WNT, Warszawa 2009.

3. Makosza M., Fedoryński M., Podstawy syntezy organicznej. Reakcje jonowe i rodnikowe, Oficyna Wydawnicza Politechniki Warszawskiej, Warszawa 2006.

4. Skarżewski J., Wprowadzenie do syntezy organicznej, PWN, Łódź 1999.

5. Buza D., Sas W., Szczeciński P., Chemia organiczna. Kurs podstawowy, Oficyna Wydawnicza Politechniki Warszawskiej, Warszawa 2006

Additional bibliography:

1. Willis C., Wills M, Synteza organiczna, Wydawnictwo Uniwersytetu Jagiellońskiego, Kraków 2004.

2. Smith M.B., March J., Advanced Organic Chemistry, Reaction, Mechanism and Structure, J.Wiley & Sons, New Jersey 2007.

3. Vogel A.I., Preparatyka organiczna, WNT, Warszawa 2006.

4. Przewodnik do nomenklatury związków organicznych, Polskie Towarzystwo Chemiczne, Warszawa 1994.

Result of average student's workload

Activity		Time (working hours)
1. Participation in lectures, exercises		45
2. Participation in the laboratory	45	
3. Participation in consultations related to the preparation of present	8	
synthesis and preparation of the student for classes laboratory tests	, test summary and exam	5
4. Preparation of the presentation		7
5. The research literature and preparing the report after class laboration	itory	5
6. Preparation for final test		7
7. Preparation for the test carried out on exercises	17	
8. Preparation for the exam and the presence of the exam		
Student's wo	rkload	
Source of workload	hours	ECTS
Total workload	139	5
Contact hours	101	3
Practical activities	53	2